TOTAL SYNTHESIS OF THE CHIRAL LACTONE DERIVED FROM THERMOZYMOCIDIN (MYRIOCIN). RELATIVE CONFIGURATION OF THE NATURAL PRODUCT.

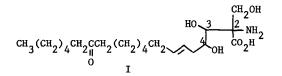
By C. H. Kuo and N. L. Wendler

Merck Sharp & Dohme Research Laboratories, Division of Merck & Co., Inc.

P. O. Box 2000, Rahway, New Jersey 07065

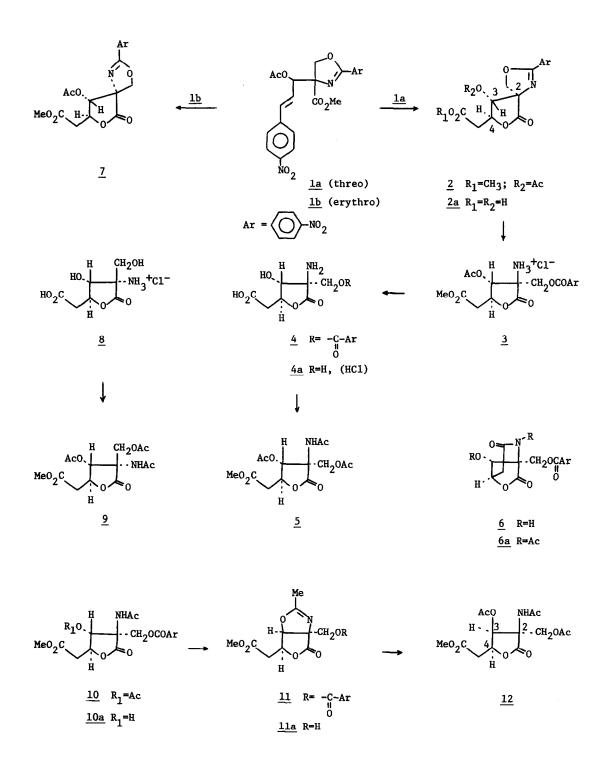
(Received in USA 17 October 1977; received in UK for publication 21 November 1977)

The antibiotic thermozymocidin (myriocin) was isolated in 1972 from a thermophilic fungus and assigned structure I, exclusive of its stereochemistry.¹ In 1973 Bagli <u>et al</u>.² published



a definitive article on the stereochemistry of this substance in which the relative configurational assignments were based on observations made <u>inter al</u>. with the chiral lactonic ester derived from the natural product. Our interest in the total synthesis of this antibiotic directed our attention to the preparation of the chiral lactonic ester and its subsequent conversion to the natural product. In the course of our investigation we have synthesized the pertinent isomeric systems and have established that the structure of the chiral lactonic ester derived from thermozymocidin does not conform to the functionally³ <u>all-trans</u> configuration <u>5</u> proposed by Bagli <u>et al</u>. (<u>loc</u>. <u>cit</u>.). The natural product, in fact, possesses the relative configuration represented by the functionally³ <u>all-cis</u> structure <u>12</u>.

Condensation of 2-p-nitrophenyl-4-carbomethoxy-2-oxazoline⁴ with p-nitrocinnamaldehyde (THF-DBN/25°) afforded the two diastereoisomeric aldols (1:1, 80%) separable by differential solubility in chloroform into the <u>threo</u> isomer mp 195-198°, <u>Acetate 1a</u>⁵ mp 203-205° and the <u>erythro</u> isomer mp 160-161°, <u>Acetate 1b</u>⁵ mp 150-151°. Ozonolysis $[0_3/CH_3OH-CH_2Cl_2/-78°; (CH_3)_2S]$ of <u>1a</u> followed by purification (silica gel/6% Acet-Chf./0°) and condensation of the unstable intermediate aldehyde (mp 126-128°; ir (CHCl_3) 5.7, 6.09, 6.22, 6.51, 7.42, 11.51, 11.70µ) with methyl acetate anion (CH₃COOCH₃/(Me_3Si)_2NLi-THF/-78°/20 min) proceeded almost exclusively <u>via</u> a <u>trans</u>-oriented pathway to give the <u>Oxazoline lactone 2</u>⁵ (40%) mp 166-168°; ir (CHCl_3) 5.58, 5.72, 6.09, 6.2, 6.54, 7.46, 11.52, 11.71µ). The <u>trans</u>-orientation: C₃-OAc/C₄-CH₂CO₂CH₃ was ascertained by saponification of <u>2</u> (2 eq KOH-CH₃OH/25°) to the <u>non-lactonizable</u> hydroxy acid <u>2a</u> which reverted to <u>2</u> on acetylation followed by methylation (CH₂N₂/THF-Et₂O). The <u>cis</u>-



orientation: C_2 -NHAc/ C_4 -CH₂CO₂CH₃ was determined by two-stage hydrolysis $2 \rightarrow 3$ (2.5N HC1-CH₃OH/25°) followed by $3 \rightarrow 4 + 4a$ (1:1) (2.5N HC1/90°); subsequent treatment of 4 with DCC in pyridine at 25° gave the <u>Bicyclic lactam 65</u> (mp 220-223°, ir (Nujol) 2.79, 5.60, 5.75, 5.91µ). Acetylation of 4 alternatively produced the corresponding O,N-diacylated lactam $6a^5$ (mp 182-184°; M⁺ 420, 348, 291; ir (CHC1₃) 5.59, 5.70, 5.92, 6.59, 7.49µ; uv (CH₂C1₂) 259nm (ε , 13,400). The functionally³ <u>all-trans</u> stereochemistry of this series was thereby established. Esterification of 4a followed by acetylation⁶ provided the corresponding <u>Lactonic ester 55</u> (mp 136-138°; ir (CHC1₃) 2.91, 5.59, 5.69 (sh), 5.7, 5.94µ; M⁺ 345; nmr (CDC1₃) δ 2.05, 2.12 (9H, 2s, 0 -0-C-CH₃ + NHCOCH₃), 3.02 (2H, broad d, J=6.5 Hz, -CH₂-C-), 3.72 (3H, s, COCH₃), 4.23, 4.57 (2H, ABq, J=11 Hz, -CH₂-O-) 4.73 (1H, m, -CH₂-CH-O-), 5.67 (1H, d, J=6.5 Hz, AcO-C-H), 6.45 (1H, broad s, NH-COCH₃) which differed from the lactonic ester derived from thermozymocidin² in the nmr by the splitting pattern of the -CH₂OAc group.

In a parallel reaction sequence the <u>erythro</u> isomer <u>1b</u> was transformed to the <u>Oxazoline</u> <u>lactone</u> 7^5 (mp 197-200°; ir (CHCl₃) 5.58, 5.72, 6.09, 6.26, 6.56, 7.48, 11.05, 11.72µ) in which the stereochemistry at C_2/C_3 follows by difference with its isomer <u>2</u>. Further, total hydrolysis of <u>7</u> (cf $2 \rightarrow 3 \rightarrow 4a$) yielded <u>8</u> which on acetylation and methylation (CH₂N₂-THF-Et₂O) gave the same <u>Lactonic ester</u> 9^5 [mp 176-178°; ir (CHCl₃) 2.91, 5.59, 5.71, 5.91µ; ms M⁺ 345, 314, 302, 285, 272; nmr (CDCl₃) 2.02, 2.08, 2.13 (9H, 3s, 0-C-CH₃ + NHCOCH₃), 2.90 (2H, d, J=6 Hz, -CH₂-C-), 3.70 (3H, s, COOCH₃), 4.50 (2H, s, -CH₂-O-), 4.95 (1H, q, -CH₂-C-O-), 5.23 (1H, d, <u>H</u> J=6 Hz, AcO-C-H), 6.51 (1H, broad s, -NH-COCH₃)] as that derived from <u>8</u> by initial esterification and subsequent acetylation.⁶ This result established the <u>trans</u>-relationship of the C₄-CH₂CO₂CH₃ side-chain with respect to both the C₃-OH and the C₂-NH₂ functions since otherwise lactonization or lactamization would have occurred. The lactonic ester <u>9</u> differed from that derived from thermozymocidin² in chemical shifts of the acetoxy methylene and methinyl protons.

The essential identity of the splitting pattern of the acetoxy methylene group of <u>9</u> in the nmr with that of the natural lactonic ester suggested the same C_2/C_3 but different C_3/C_4 configurations in the two substances. Consequently the C_3 -OH of isomer <u>5</u> was inverted. To this end acetylation⁶ of <u>3</u> yielded <u>10</u>⁵ (mp 95-97°) converted (CH₃OH-HCl) to <u>10a</u>⁵ (mp 161-163°). Treatment of <u>10a</u> with trifluoromethanesulfonic anhydride in pyridine at 0° proceeded with accompanying inversion at C_3 and formation of the oxazoline <u>11</u>⁵ (mp 183-185°; ir (CHCl₃) 5.59, 5.75, 6.02µ; M⁺ 392). Removal of the p-nitrobenzoyl group (CH₃OH-KOAc) followed by hydrolysis (2.5N HCl/CH₃OH-THF/25°) completed the N \rightarrow 0 acetyl transfer yielding the corresponding Oacetyl amine hydrochloride. Acetylation of the latter yielded the functionally³ <u>all-cis</u> <u>Lactonic ester 12</u>⁵ (mp 142-144°); ir (CHCl₃) 2.91, 5.59, 5.71, 5.92µ; ms M⁺ 345, 302, 272; nmr (CDCl₃) 2.03, 2.04 (6H, 2s, -0-C-CH₃), 2.14 (3H, s, NH-COCH₃), 2.87 (2H, d, J=7 Hz, -CH₂-C-O), <u>H</u> 5.73 (1H, d, J=5.5 Hz, AcO-C/C-H), 6.02 (1H, broad s, NHCOCH₃) identical spectroscopically with the chiral lactonic ester derived from thermozymocidin.²,⁷

213

Acknowledgment. The authors are greatly indebted to Dr. A. Douglas and Mr. R. Reamer for nmr spectra and discussions.

References

- (a) F. Aragozzini, P. L. Manachini, R. Craveri, B. Rindone and C. Scholastico, <u>Tetrahedron</u>, <u>28</u>, 5493 (1972); (b) D. Kluepfel, J. F. Bagli, H. Baker, M. Charest, <u>A. Kudelski</u>, S. N. Sehgal and C. Vezina, <u>J. Antibiotics</u>, <u>25</u>, 109 (1972).
- 2. J. F. Bagli, D. Kluepfel and M. St. Jacques, J. Org. Chem., 38, 1253 (1973).
- 3. The term "functionally" refers to functions: C_2 -NH₂, C_3 -OH, and C_4 -CH₂CO₂CH₃ which are transformationally involved in the configurational determinations.
- 4. S. Ginsburg and I. B. Wilson, J. Am. Chem. Soc., 86, 4716 (1964).
- 5. Satisfactory elemental analyses and acceptable nmr spectra were obtained for all pertinent compounds.
- Acetylations were effected with acetic anhydride in pyridine (1:3/25°) and esterification with methanol saturated with hydrogen chloride at 0-25°.
- 7. The same stereochemistry has recently been derived from X-ray determination on thermozymocidin. (Private communication from Dr. C. Scholastico of the Inst. Chim. Org. Centro Naz. Chim. Sost. Org. Nat. CNR, Milano. We thank Dr. Scholastico for informing us of this finding prior to its publication.)